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Molecular modeling of alkaloids bouchardatine and orirenierine binding to sirtuin-1 (SIRT1)

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Objective Bouchardatine (1) is a β -indoloquinazoline alkaloid isolated from the plant *Bouchardatia neurococca*, acting as a modulator of adipogenesis and lipogenesis, and as an anticancer agent. The natural product functions as an activator of proteins adenosine 5'-monophosphate (AMP)-activated protein kinase (AMPK) and sirtuin 1 (SIRT1). We used molecular modeling to investigate the SIRT1-binding capacity of compound 1 and various structural analogues, such as orirenierine A (2) and orirenierine B (3) isolated from the medicinal plant *Oricia renieri*.

Methods We investigated the binding to human SIRT1 (hSIRT1) of 25 natural products including the β -indoloquinazoline alkaloids $\mathbf{1}-\mathbf{3}$ and analogues, in comparison with the reference product sirtinol (R and S isomers). A sirtinol binding model was elaborated starting from the closed and open state conformations of the catalytic domain of hSIRT1 (PDB structures 4KXQ and 4IG9). For each compound bound to SIRT1, the empirical energy of interaction (Δ E) was calculated and compared to that of sirtinol.

Results In our model, compound 1 was found to bind modestly to the sirtinol site of SIRT1. In contrast, the presence of a phenolic OH group at position 7 on the quinazolinone moiety conferred a much higher binding capacity. Compound 2 provided SIRT1 protein complexes as stable as those observed with sirtinol. The replacement of the hydroxy substituent (2) with a methoxy group (3) reduced the SIRT1 binding capacity. Other SIRT1-binding natural products were identified, such as the alkaloids orisuaveolines A and B. Structure-binding relationships were discussed.

Conclusion The study underlines the capacity of β -indoloquinazoline alkaloids to interact with SIRT1. This deacetylase enzyme could represent a molecular target for the alkaloid **2**. This compound merits further attention for the design of drugs active against SIRT1-dependent pathologies.

1 Introduction

The alkaloid bouchardatine (1) was isolated twenty years ago from the aerial parts of the plant *Bouchardatia*

neurococca (F. Muell.) Baill. (Rutaceae) collected in the state forest of Polmaily (Queensland, Australia). It is a small rainforest tree commonly known as "union nut" that is endemic to eastern Australia. The name

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Bouchardatia was given after the French pharmacist Apollinaire Bouchardat (1806 - 1886). The term neurococca derives from the Greek "neuron" (nerve) and "coccos" (a berry), referring to the ribbed carpels or cocci. There is no mention of the use of this plant in traditional medicine, unlike other related Rutaceae species. Chinese medicinal materials originated from Rutaceae family are frequently used to treat eczema, rheumatism, animal bites, and other health problems [1, 2]. Diverse bioactive natural products have been isolated from Bouchardatia neurococca, including limonoids such as veprisonic and isoveprisonic acid, and alkaloids such as bouchardatine, dictamnine, rutaecarpine and their derivatives [3]. Sesquiterpenes have been isolated from the plant leaf oil so far, while the phytochemical content of this species has been little investigated [4].

Bouchardatine (1) comprises a quinazolin-4-one core linked to an indole unit equipped with a formyl group (Figure 1). The alkaloid can be isolated from a methanol extract of the bark of Bouchardatia neurococca after chromatography. Alternatively, total syntheses of the compound have been described [5]. An efficient 5-step synthesis from anthranilamide afforded gram-quantities of compound 1 for pharmacological studies [6]. The product has revealed interesting properties in treating metabolic diseases and cancer. A lipid-lowering effect was initially described using cultured 3T3-L1 adipocytes. Compound 1 was found to reduce dose-dependently the proliferation of adipocytes and the accumulation of lipids by suppressing the expression of adipogenic factors such as peroxisome proliferator-activated receptor y (PPARy) and CCAAT enhancer-binding proteins (C/EBP) $\alpha/\beta/\delta$. The lipid-lowering effect of compound 1 was linked to an activation of the adenosine 5'-monophosphate (AMP)-activated protein kinase (AMPK) pathway [6]. Further studies on the effects of compound 1 on adipogenesis and lipogenesis led to a better understanding of the mechanism of action of the natural product. RAO et al. [7] found that compound 1 increased the activity of the protein sirtuin 1 (SIRT1) to facilitate activation of AMPK by the enzyme liver kinase B1 (LKB1). Activation of the SIRT1-LKB1-AMPK signaling pathway by compound 1 in adipose tissue and liver is at the origin of the anti-obesity effect. The capacity of compound 1 to reduce obesity was demonstrated using a specific high-fat diet model in mice [7].

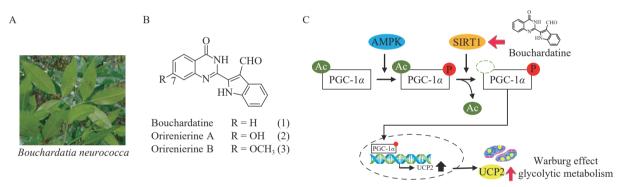


Figure 1 Bouchardatine, plant origin, and mechanism of action

A, the aerial part of the plant Bouchardatia neurococca (F. Muell.) Baill. (https://alphitonia.com/EditSpeciesE.cshtml?id = 555). B, structures of the alkaloids bouchardatine (1), orirenierine A (2), and orirenierine B (3). C, a schematic illustration of the mechanism of compound 1 as an activator of the deacetylase SIRT1 to upregulate the expression of uncoupling protein 2 (UCP2) implicated in the glycolytic metabolism of cancer cells.

The interest towards compound 1 has been underlined in a recent study of the anticancer activity of the product in a murine model for rectal cancer [8]. The growth of a xenografted HCT-116 tumor in mice was reduced markedly upon daily administration of compound 1 at the dose of 50 mg/kg (i.p. injection). More importantly, the authors demonstrated that compound 1 activated the metabolic regulator SIRT1, leading to the subsequent activation of peroxisome proliferators activated receptor γ coactivator 1α (PGC- 1α), coupled to a downstream upregulation of uncoupling protein 2 (UCP2) and an alteration of the cancer cell metabolism, as depicted in Figure 1 [8]. UCP2 is an antioxidant mitochondrial protein involved in the maintenance of the Warburg effect in

cancer cells. Most importantly, this study demonstrated that activation of the SIRT1-PGC1α-UCP2 axis by compound 1 was at the origin of the anticancer effect. A mutation (H355A) in the enzymatic site of SIRT1 abolished the stimulation effect of compound 1, suggesting that this protein could be a direct target for the natural product [8].

Based on these considerations, we attempted to evaluate the capacity of compound 1 to interact with SIRT1 using molecular modeling. The protein functions as a nicotinamide adenine dinucleotide (NAD+)-dependent class III histone deacetylase (HDAC). The tridimensional structure of the catalytic domain of human SIRT1 has been solved by X-ray crystallography [9]. We used the HDAC domain of human SIRT1 (hSIRT1, PDB: 4KXQ and 4IG9) to perform a docking study to locate the potential binding site for compound **1**, using the known hSIRT1 inhibitor sirtinol as a reference. The same model has been used successfully by different research groups to identify new inhibitors of hSIRT1 [10, 11]. An *in silico* model of the bouchardatine-hSIRT1 complex was elaborated, and then used to compare the binding capacity of over 20 alkaloids, including orirenierine A and B (compounds **2** and **3**, respectively), orisuaveoline A and B, luotonins A – F, and other products.

2 Materials and methods

2.1 Molecular structures and software

Two tridimensional structures of hSIRT1 in a closed (4KXQ) and an open (4IG9) state conformation were retrieved from the Protein Data Bank (www.rcsb.org) ^[9]. Docking experiments were performed using the GOLD software (GOLD 5.3 release, Cambridge Crystallographic Data Centre, UK). Before starting the docking procedure, the structure of the ligands had been optimized using a classical Monte Carlo (MC) conformational searching procedure as described in the BOSS software ^[12]. Potential GSK-3 β -binding sites for the different products were searched using the web server Computed Atlas of Surface Topography of proteins (CASTp) 3.0, and visualized with the molecular modeling software Chimera 1.15 ^[13, 14].

2.2 In silico molecular docking procedure

With each structure, the drug binding site and surrounding amino acids were defined based on shape complementarity criteria. Shape complementarity and geometry considerations were in favor of a docking grid centered in the volume defined by the central amino acid. Within the binding site, side chains of specific amino acids were considered fully flexible. For structure 4KXQ, the flexible amino acids were Phe273, Phe297, Ile316, Tyr317, Ile347, His363, Ser365, Phe366, Val412, and Phe413. The ligand was always defined as flexible during the docking procedure. Up to 100 poses that were energetically reasonable were kept while searching for the correct binding mode of the ligand. The decision to keep a trial pose was based on ranked poses, using the Picewise Liner Potential (PLP) fitness scoring function (which is the default in GOLD version 5.3 used here) [15]. The same procedure was used to establish molecular models for the various natural products (Table 1) and the reference compounds. The empirical potential energy of interaction (ΔE) for the ranked complexes was evaluated using the simple expression ΔE (interaction) = E (complex) – [E (protein) + E (ligand)]. For that purpose, the Spectroscopic Empirical Potential Energy function Spectroscopic Potential Algorithm for SImulating Biomolecular conformational Adaptability

(SPASIBA) and the corresponding parameters were used. Free energies of hydration (ΔG) were estimated using the Molecular Mechanics/Generalized Born Surface Area (MM/GBSA) model in MC simulations within the BOSS software [16]. The stability of the receptor-ligand complex was evaluated based on the empirical potential energy of interaction [17, 18]. The MM/GBSA procedure was used to evaluate free energies of hydration (within the BOSS program) [12] in relation to aqueous solubility. Molecular graphics were created and analyzed using Discovery Studio Visualizer, Biovia 2020 (Dassault Systèmes BIOVIA Discovery Studio Visualizer 2020, San Diego, Dassault Systèmes, 2020).

Table 1 Calculated potential energy of interaction (ΔE , kcal/mol) and free energy of hydration (ΔG_{hyd} , kcal/mol) for the interaction of indoloquinazolinone alkaloids (and selected natural products) with SIRT1 (PDB: 4KXQ)

| Compound | CIDa | ΔE | ΔG_{hyd} |
|-------------------|-----------|---------|------------------|
| (R)-Sirtinol | 1376646 | - 75.70 | - 32.60 |
| (S)-Sirtinol | 1376645 | - 74.50 | - 29.60 |
| Bouchardatine | 135800540 | - 49.50 | -21.70 |
| Orirenierine A | _ | -71.40 | -23.70 |
| Orirenierine B | _ | - 56.90 | - 25.40 |
| Orisuaveoline A | 135953334 | - 65.40 | -18.70 |
| Orisuaveoline B | 25157868 | - 64.40 | - 26.15 |
| Ketoyobirine | 275187 | - 62.90 | - 20.70 |
| Hortiacine | 378227 | - 54.00 | - 18.70 |
| Evodiamine | 442088 | - 52.30 | - 17.30 |
| Hydroxyevodiamine | 56967381 | - 63.90 | - 18.80 |
| Angustine | 441983 | - 58.40 | - 1.40 |
| Angustoline | 3084765 | -48.00 | - 3.40 |
| Naucletine | 5320037 | - 49.40 | - 21.90 |
| Rhetsinine | 99652 | - 54.20 | - 17.30 |
| Tephcalostan | 11013898 | - 59.10 | -25.90 |
| Tephcalostan B | 25243039 | - 56.65 | - 22.15 |
| Tephcalostan C | _ | - 58.80 | - 27.80 |
| Tephcalostan D | 102283894 | - 63.10 | - 21.80 |
| Luotonin A | 10334120 | - 48.95 | - 18.50 |
| Luotonin B | 10017730 | - 42.55 | - 12.60 |
| Luotonin C | 102369825 | - 44.80 | - 19.10 |
| Luotonin D | 102369826 | - 55.30 | - 17.00 |
| Luotonin E | 11483874 | - 45.95 | - 23.20 |
| Luotonin F | 13545779 | - 53.55 | - 13.10 |

^aCompound IDentity number, as defined in PubChem (https://pubchem.ncbi.nlm.nih.gov). "–" represents that these compounds are not listed in the PubChem databank.

2.3 MC simulations

The two most widely used methods to investigate protein-ligand stability and affinity are Molecular Dynamics (MD) and MC simulations. MD simulation of proteins is a challenge that requires careful consideration of the 1st law of thermodynamics [19]. Both methods employ an empirical force field to control the total energy (MC, energy minimization) and forces (MD, Newton equations of motion). To use MD simulations confidently, a force field parameterized for dynamical properties is required. The development of a reliable and accurate force field for conformational analysis remains a concern. It requires accuracy of the force field over the whole potential surface, rather than in the region of the global minimum [20]. The most used academic force fields, such as Chemistry at HARvard Macromolecular Mechanics (CHARMM), Assisted Model Building with Energy Refinement (AMBER), and GROningen MOlecular Simulation (GROMOS), do not exhibit the required vibrational spectroscopic quality. Minimization of a protein structure with normal modes can result in the calculation of imaginary wavenumbers corresponding to maxima in the potential energy (transition states, mainly due to inadequate barriers to internal rotation). The spectroscopic SPASIBA force field has been specifically developed to provide refined empirical molecular mechanics force field parameters, as described in other studies [17, 21]. For this reason, we adopted MC simulations rather than MD, which requires a substantial increase in computing time to achieve the same level of convergence [22].

3 Results

3.1 Construction of the sirtinol-SIRT1 binding model

We started our analysis with two structural models of SIRT1. The PDB structures 4KXQ and 4IG9 refer to a closed and an open state conformation of the catalytic domain of hSIRT1, respectively [9]. The two structures were very similar, as shown from the superimposed model in Figure 2A. The position of the adenosine 5'-diphosphoribose (ADP-ribose) ligand was identical for the two structures; the secondary structure of the binding site was preserved. The only significant difference was around the loop Phe273-Ile279, which seemed to act as a gate, closing or opening the site, as represented in Figure 2B, though this gate did not affect sirtinol binding around site I347. To locate the drug binding site, we used the web server CASTp 3.0, which was specifically designed for the analysis of protein topography. This tool is mostly used in identifying drug binding sites and quantifying the volume of the binding pocket [14]. A view of the drug-binding pocket defined with CASTp using the open state conformation (4IJ9) is shown in Figure 2B. Two potential binding sites were identified for the reference inhibitor sirtinol. The best site was located near residue Ile347 $(\Delta E = -71.30 \text{ kcal/mol})$ and $\Delta G = -30.40 \text{ kcal/mol})$ and a secondary site near residue Ile316 ($\Delta E = -55.40 \text{ kcal/mol}$ and $\Delta G = -20.00$ kcal/mol). The main sirtinol binding site Ile347 was juxtaposed to the ADP-ribose binding site

(Figure 2C). This site Ile347 was defined as the potential binding site for the other ligands tested in our docking analysis, and was juxtaposed to the C-pocket, where the nicotinamide moiety of NAD+ binds and the hydrolysis takes place. Drug binding to this site is known to block the transformation of NAD+to productive conformation and to inhibit the deacetylase activity [23, 24]. The secondary weak site (Ile316) has never been mentioned in other studies; it is considered too weak and irrelevant. Our analysis was thus focused on the primary site.

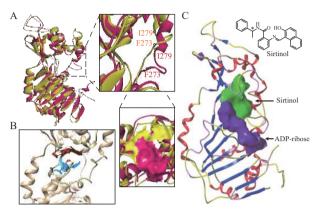


Figure 2 Molecular model of sirtinol bound to SIRT1 A, superimposition of the open state (4IJ9, magenta) and closed state (4KXQ, yellow) conformations of SIRT1. A closed view of the ADP-ribose binding zone is shown (inset) to underline the different positions of the F273-I279 flexible loop, closed or opened. B, a view of the sirtinol binding zone (as defined by CASTp 3.0) delimited by the hydrophilic (blue) and hydrophobic (red) areas. C, global view of the catalytic domain of hSIRT1 with the ADP-ribose (violet) and sirtinol (green) binding regions highlighted. The structure of sirtinol is shown (with the asymmetry center *).

We defined the pocket around residue Ile347 as the sirtinol binding site and compared the two isomers (R)sirtinol and (S)-sirtinol, but no significant difference was found between the two products: both produced stable complexes with SIRT1 as indicated from the ΔE values collated in Table 1. As expected, there is no enantioselective inhibitory effect toward SIRT1. It has been shown that (R)- and (S)-sirtinol have similar inhibitory effects on the yeast and human SIRT1 enzymes [25]. Altogether, our sirtinol-SIRT1 model falls in good agreement with models previously published. We kept the model (4KXQ) for the subsequent docking analysis.

3.2 Interaction of bouchardatine with SIRT1

A docking model of compound 1 bound to SIRT1 was elaborated, as shown in Figure 3, where compound 1 sits in the small cavity close to the ADP-ribose binding site. Several interactions stabilized the drug-protein complex, including a π -stacking interaction with Phe273 and an Hbond between NH-indole and Ile347. Additional van der Waals contacts contributed to maintaining

drug-protein interaction but overall, the SIRT1-1 complex was not as stable as the one formed between SIRT1 and sirtinol. The calculated ΔE and ΔG were weaker with compound 1 versus sirtinol (Table 1). The ΔE value reached – 75.70 kcal/mol with (R)-sirtinol compared to only – 49.50 kcal/mol with compound 1, a loss of 34% in binding energy. Under these conditions, the alkaloid can be considered as a potential weak binder.

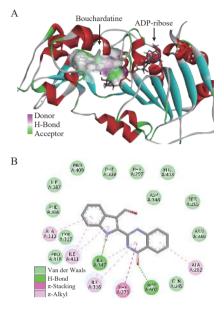


Figure 3 Molecular model of bouchardatine (1) bound to SIRT1 (PDB: 4KXQ)

A, SIRT1-1 complex in the presence of ADP-ribose, with the corresponding H-bond donor/acceptor elements. The protein is shown with the α -helices (red) and β -sheets (cyan). B, binding map contacts for 1 bound to SIRT1 (color codes indicated).

3.3 Interaction of orirenierines A and B with SIRT1

Next, we investigated the binding of bouchardatine derivatives with the protein. The closest naturally-occurring analogues known are the alkaloids orirenierine A (2) and orirenierine B (3), which have been isolated from the Cameroonian medicinal plant *Oricia renieri* (Rutaceae) [26]. They only differ from compound 1 by the presence of a hydroxy (2) or methoxy (3) substituent at position 7 on the quinazolinone heterocycle (Figure 1). The plant *Oricia renieri* is used in traditional medicine to treat inflammatory conditions, such as mastitis [27]. Compound 2 was found to display a higher anti-inflammatory potential compared to compounds 3 and 1 in a blood-based test of zymosan-induced neutrophil activation, showing half maximal inhibitory concentration (IC₅₀) values of 2.7, 3.8, and 4.9 μ mol, respectively [26].

The SIRT1 docking analysis performed with compounds ${\bf 1}-{\bf 3}$ yielded useful information. The alkaloid ${\bf 2}$ equipped with a 7-hydroxyl group can form very stable complexes with SIRT1, as judged from the calculated ΔE values (Table 1). In this case, the ΔE value reached

- 71.40 kcal/mol, a value close to that calculated with the reference product sirtinol. This product displays anti-inflammatory and anticancer effects [28, 29]. A model of the SIRT1-2 complex is presented in Figure 4. The π -stacking interaction between the quinazolinone and Phe273 was preserved, but in addition, there were two potential Hbonds implicating the hydroxyl group on the drug: one with Tyr280 and one directly with the ADP-ribose ligand (Figure 4A). The hydroxyl group pointed toward the ADPribose site in a solvent-accessible area (Figure 4B). The capping of this OH group with methyl reduced the binding interaction. The methoxy analogue 3 was much less prone to binding to SIRT1 than compound 2 ($\Delta E =$ - 56.90 and - 71.40 kcal/mol, respectively). Compound 2 bound to the same site as sirtinol but the orientation of the compound within the site was significantly different (Figure 5A). The 7-OH group of compound 2 anchored

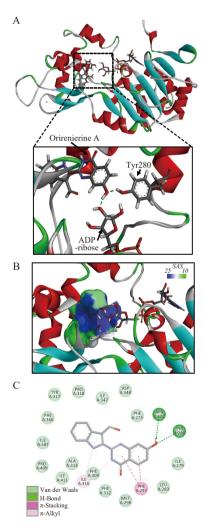


Figure 4 Molecular model of orirenierine A bound to SIRT1

A, 1-SIRT1 binding complex. The close-up view shows the two bifurcated H-bonds between the phenolic OH of the drug and residue Tyrosine 280 (Y280) of SIRT1 and the terminal ribose unit of ADP-ribose. B, a detailed view of the drug-binding site with the solvent accessible surface (SAS) colored. C, binding map contacts orirenierine A bound to SIRT1 (color codes indicated).

the molecule to ADP-ribose whereas the 7-OCH₃ group of compound 3 pulled apart the molecule from the ADPribose (Figure 5B).

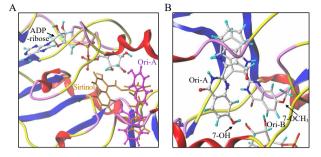


Figure 5 Compounds orientation in the binding site A, detailed view of the drug-binding pocket of SIRT1 showing the distinct orientation of orirenierine A (Ori-A) and sirtinol. The two compounds are facing the ADP-ribose ligand. B, superimposed view of orirenierine A (Ori-A) and orirenierine B (Ori-B) bound to SIRT1. Note the distinct orientations of the 7-hydroxy (Ori-A) and 7-methoxy group (Ori-B) (arrows).

3.4 Other alkaloids

The identification of orirenierine A (2) as a potential SIRT1 binder prompted us to look for other indoloquinazolinone alkaloids susceptible to forming stable complexes with the protein. We identified a dozen natural products bearing an indoloquinazolinone unit and extended the search to 20 compounds, including quinazolinocarboline alkaloids like evodiamine, coumestan derivatives like tephcalostans, and pyrrologuinolinequinoline alkaloids like luotonin A and its derivatives (Figure 6). However, in most cases these compounds were contained in an extended pentacyclic core system, quite distinct from the bi-unit system of compounds 1 - 3. For example, the two alkaloids orisuaveolines A and B isolated from the plant *Oricia suaveolens* [30] bore the same β indologuinazoline core as compounds 1 - 3 but the guinazoline and indole unit were fused to form a penta (orisuaveoline A) or hexa (orisuaveoline B) cyclic rigid system. SIRT1 docking analysis was performed with all compounds (Table 1). We did not identify a natural product superior to orirenierine A for binding to SIRT1. However, useful information was obtained.

Orisuaveolines A and B bore the same -OH or -OCH₃ as orirenierine A (2) and orirenierine B (3), respectively. These two compounds were superior to compound 1, thus providing another piece of evidence for the key role of the phenolic OH group in protein binding. However, the rigid polyaromatic system was apparently less favorable than the bis-unit of compounds 1 - 3. The ΔE value calculated with orisuaveoline A was only 10% higher than that of orirenierine A. The anti-inflammatory action of this compound would be worth investigating, which has not been studied thus far.

The quinolone alkaloid evodiamine (isolated from the fruit of Evodia rutaecarpa and other plants) displayed a large range of biological activities, including antitumor, anti-inflammation, and antimicrobial effects [31, 32]. It represented a lead structure for the design of anticancer compounds [33], with its anticancer activity being considered as SIRT1-mediated [34]. SIRT1 was likely to play a significant role in the capacity of evodiamine to inhibit the migration and invasion of cancer cells [35], but our calculations suggest that SIRT1 is not a direct binding target for evodiamine. At least, we can underline that the compound cannot form highly stable complexes upon binding to the sirtinol site of the protein. However, it is

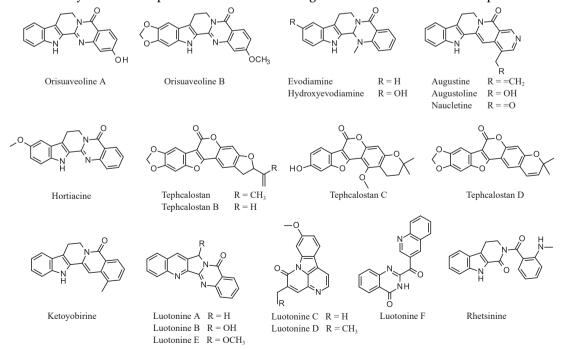


Figure 6 Structure of the other natural products evaluated as potential SIRT1 binders (as indicated in Table 1)

interesting to note that 10-hydroxyevodiamine demonstrated a better SIRT1 binding capacity than that of evodiamine (Table 1). The OH-group significantly reinforced the stability of the alkaloid-protein complex, with a gain of ΔE value of about 15% (ΔE = - 52.30 and - 63.90 kcal/mol for evodiamine and hydroxyevodiamine, respectively). This is interesting as 10-hydroxyevodiamine has been used to design fluorescent probes and tumor-active water-soluble derivatives [36, 37]. The 10-OH group can be exploited to reinforce evodiamine binding to SIRT1, whereas the N-methyl group is apparently unessential. The little-known alkaloid ketoyobirine, which is rarely investigated [38], is as efficient as evodiamine in binding to SIRT1. In short, several compounds could be tested but only one stands out as a promising SIRT1 binder: orirenierine A.

4 Discussion

More than 150 naturally occurring quinazolinone alkaloids have been identified by now [39, 40]. Some of them have been thoroughly investigated, such as the luotonins [41], while others have received little attention, such as the quinazolinocarboline alkaloid hortiacine [42]. This is also the case for the alkaloids orirenierines A and B, and orisuaveolines A and B isolated from Oricia renieri and Oricia suaveolens, respectively, by the same research group [26, 30]. These alkaloids have been described but their mechanisms of action remain unknown. In contrast, the quinazolinone alkaloid bouchardatine (1) has received attention owing to its potential in the treatment of obesity-related metabolic disorders and cancers. The signaling pathways modulated by this natural product have been characterized, notably the SIRT1-PGC1 α -UCP2 axis, and a recent study suggested that SIRT1 could be a direct target for the natural product [8]. Based on this study, and considering the structural analogy between compound 1 and orirenierines A and B (2 and 3), we established a docking study to investigate this topic.

The analysis indicated the superior capacity of the related alkaloid compound 2 compared to compound 1 to bind to hSIRT1. This alkaloid 2, which is structurally close to compound 1 (Figure 1), showed better adaptability to form stable complexes with hSIRT1 than compound 3 and all the other polycyclic indologuinazolinone alkaloids tested. The observation is of great importance from a medicinal point of view because compound 2 derives from the Cameroonian plant Oricia renieri (Rutaceae) and is used locally to treat different types of infections and cancers [26, 30]. Our analysis suggests that SIRT1 may effectively represent a protein target for these alkaloids. They have the capacity to bind to the sirtinol-binding site, in the pocket facing the ADP-ribose ligand. Compound 1 emerged as a relatively weak SIRT1 binder compared to the reference sirtinol, whereas compound 2 is

quasi-equipotent to sirtinol, at least in terms of binding energy (ΔE). The phenolic OH which distinguishes compound 1 from 2 is a prime element for protein binding, and the capping of this group in the form of a methoxy (compound 3) reduces the binding interaction. This OH group is undoubtedly a key protein binding element. It is noteworthy that the potency of the three compounds in terms of SIRT1 binding is comparable to their potency as cytotoxic agents against cancer cells. Compound 2 was found to be more cytotoxic toward PC3 prostate cancer cells compared with compounds 3 and 1, with IC₅₀ values of 23.2, 36.5, and 39.4 µmol, respectively [26]. In other words, the higher capacity of 2 to interact with SIRT1 correlates with its higher cytotoxic potential. This compound merits further attention. It is an important component of extracts of Oricia renieri [27], and for the first time, a molecular target for it is proposed.

SIRT1 binders were actively searched because these compounds showed potential in the treatment of cancer and other human diseases, including autoimmune diseases [43-45]. It is therefore interesting to propose a new scaffold β -indoloquinazoline for the design of SIRT1-interacting drugs. This scaffold has been little investigated thus far. However, there are two noticeable exceptions. Recently, the drug IQZ23 has been proposed as a candidate for the treatment of obesity and obesity-related metabolic disorders. This β -indologuinazoline derivative was found to activate the AMPK pathway but its molecular target is not precisely known [46]. The preclinical development of IQZ23 has been initiated [47]. It would be interesting to determine its capacity to modulate the activity of SIRT1. Similarly, compounds 1 - 3 present a structural analogy with the anticancer dihydroquinazolinone derivative MHY2245 recently identified as a potent sirtuin inhibitor [48, 49]. This reinforces the idea that orirenierines can be used as templates to design SIRT1 modulators.

The present work can also guide the future design of bouchardatine analogues. Various synthetic analogues of compound 1 have been designed as inhibitors of adipogenesis/lipogenesis [50]. For example, the lipid-lowering activity of compound 1 can be considerably enhanced by replacing the quinazolin-4-one with a quinazolin-4-amine group. Potent activators of AMPK have been obtained, with compounds active *in vivo*, of interest in the treatment of metabolic diseases and non-alcoholic fatty liver disease (NAFLD) [51, 52]. Based on our observations, new compounds bearing the key 7-OH group of compound 2 could be now envisioned to reinforce SIRT1 binding and potentially their activity against different metabolic diseases.

As an important protein target, SIRT1 is associated with the activity of different Chinese medicine prescriptions, such as Taohuajing (桃花精) treating diabetic cardiomyopathy [53], Yunpi Heluo Decoction (运脾和络方) treating type 2 diabetes [54], Zhibai Dihuang Decoction (知

柏地黄汤) treating oral ulcers [55]. The identification of novel SIRT1 binders may help further to better understand the molecular basis of the bioactivity of different Chinese medicine products.

This computational study is useful to select and prioritize compounds for further laboratory experiments. Docking analysis such as those reported here have limited predictability, due to the complexity of the protein folding process and its dynamic [56]. Molecular docking represents a useful aid in the process of drug discovery and development [57, 58], but its predictive value is limited. We do not underestimate the necessity to carry out wetlab validations for the predicted SIRT1-binding candidates identified here.

5 Conclusion

In summary, SIRT1 is an NAD+-dependent deacetylase implicated in various human pathologies. Small molecule modulators of SIRT1 were searched, including SIRT1 inhibitors and activators. Here we found that alkaloids bearing an unfused β -indoloquinazoline ring system can provide potential SIRT1 binders. We characterized the interaction of the alkaloid orirenierine A (2) with human SIRT1 at the sirtinol-binding site. The specific contribution of the phenolic OH group of compound 2 was underlined. The use of various structural analogues enables the definition of structure-binding relationships.

6 Highlights

- (i) The plant alkaloid bouchardatine (1) has been found to activate the enzyme SIRT1.
- (ii) Molecular docking indicates that analogue orirenierine A (2) presents a high capacity to bind to the sirtinol site of SIRT1.
- (iii) A few other indoloquinazoline alkaloids interacting at the sirtinol site of SIRT1 have been identified.
- (iv) Structure-binding elements have been defined in the chemical series.
- (v) The β -indoloquinazoline unit can be exploited to design SIRT1-binding drugs active against cancers.

Competing interests

The authors declare no conflict of interest.

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生物碱 bouchardatine 及 orirenierine 与组蛋白去乙酰化酶 1 (SIRT1) 结合的分子建模研究

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【摘要】目的 Bouchardatine(1) 是从植物 Bouchardatia neurococca 中分离得到的一种 β-吲哚喹唑啉生物 碱,可作为脂肪生成的一种调节剂及一种抗癌药。天然产物作为蛋白腺苷 5'-单磷酸 (AMP) 活化蛋白激酶 (AMPK) 和组蛋白去乙酰化酶 1 (SIRT1) 的激活剂发挥作用。我们利用分子模型研究了化合物 1 和各种结构 类似物的 SIRT1 结合能力,如从药用植物 Oriciarenieri 中分离得到的 orirenierine A(2)和 orirenierine B(3)。 方法 我们研究了包括 β-吲哚喹唑啉生物碱 1-3 和类似物在内的 25 种天然产物与人源组蛋白去乙酰化酶 1 (hSIRT1) 的结合,并与参比产物去乙酰化酶(R和S异构体)进行了比较。从hSIRT1 催化结构域的闭合和开 放状态构象(PDB 结构: 4KXO 和 4IG9)开始阐述去乙酰化酶结合模型。对于与 SIRT1 结合的每种化合物,我们 计算了相互作用的经验能量(ΔE),并与去乙酰化酶进行比较。**结果** 在我们的模型中,发现化合物 1 与 SIRT1 的去乙酰化酶位点适度结合。相反,喹唑啉酮部分7位酚羟基表现出更高的结合能力。化合物2提供的SIRT1 蛋白复合物与去乙酰化酶观察到的一样稳定。用甲氧基(3)取代羟基取代基(2)降低了 SIRT1 的结合能力。我 们还鉴定了其他 SIRT1 结合的天然产物,如生物碱 orisuaveolines A 和 B,并讨论了结构结合关系。结论 本研 究强调了 β -吲哚喹唑啉生物碱与SIRT1相互作用的能力。这种去乙酰化酶可以代表生物碱2的分子靶标。这 种化合物在设计对 SIRT1 依赖性病理有活性的药物研究方面值得进一步关注。

【关键词】组蛋白去乙酰化酶 1;生物碱;Bouchardatine;Orirenierine;天然产物;癌症;炎症性疾病;分子建 模;药物设计