

ORIGINAL ARTICLE

COMPARISON OF HEAVY METAL LEVELS IN NATURAL SPRING AND BOTTLED DRINKING WATER IN KLANG VALLEY, MALAYSIA

Mohd Hasni J.¹, Aminnuddin M.¹, Ariza Z.¹, Azwani A.¹, Engku Nurul Syuhadah E.A.¹, Nor Asikin J.¹ and Nor Dalila M.¹¹Department of Community Health, UKM Medical Center, National University of Malaysia

ABSTRACT

A spring is a result of karsts topography where surface water has infiltrated the earth's surface recharge area, becoming part of groundwater and emerges from below to become natural spring water. From few observations, local people tend to consume this water directly for many health reasons. The objective of the study was to determine the concentration of lead (Pb) and cadmium (Cd) in natural water resources and bottled drinking water sources, and compared with the existing standard. This field assessment was carried out in 2014 as part of the educational module for public health master student. About 13 water samples were collected directly from the tubing into the pre-washed sample bottle and rinse with the sampling water in the field. Sample preservation was achieved by acidifying to pH less 4.0 with nitric acid (HNO₃). Samples were stored in a cooler with temperature between 0 to 4°C. Heavy metals were analysed by standard method for graphite furnace atomic absorption spectrophotometer (GFAAS) with Zeeman's correction. Pb and Cd content were analysed from 13 samples which consists of eight natural spring water and five flavoured bottled drinking water. The result showed that Pb content in spring water ranges between 1.8 and 37.3ppb, while the Cd content in spring water ranges between 3.0 and 23.0ppb. In the commercialised drinking water, the amount of Pb ranges between 0.4 and 2.6 ppb, while the content of Cd ranges between 0.8 and 7.0 ppb. This study indicates that there are high content of Pb and Cd in most of the natural spring water points and some bottled drinking water that are available within Klang Valley. In the absence of any specific point sources, the possibility of urban area and high traffic source leading to run off as well as rock types may result in variations observed. Hence, very worrying, especially that these sources of water were consumed directly as drinking water or eye drops without knowing its content. Further tests, coupled with supportive soil and conductivity studies, are required to test all possible similar natural sources to safeguard the health of people.

Keywords: Pb, Cd, spring natural water, drinking water, Klang Valley

INTRODUCTION

About 70% of the earth is covered with water, and only about 1% are soft water from lakes, rivers and underground sources. One type of the underground water is spring water. The forcing of the spring to the surface can be the result of a confined aquifer in which the recharge area of the spring water table rests at a higher elevation than that of the outlet.

As the water moves through the underground rocks it will dissolve the minerals and natural metals. This gives the water flavour and even carbon dioxide bubbles, depending on the nature of the geology through which it passes. Iron and tannins often give spring water an orange colour. Other minerals that usually can be found in spring water are magnesium and calcium that dissolved naturally from the rocks¹.

Two particular heavy metals of interested in this study were lead (Pb) and cadmium (Cd). This was due to the abundant presence of the two in Malaysia environment and food chain as mentioned by other studies²⁻⁶. Both are toxic to the human body as they may affect the kidney system,

cardiovascular system, and neurological system lead to various chronic diseases.

In Klang Valley there are multiple locations where it is famous for the local people to collect the spring water and consumed it without doing any form of water treatment. Either they drink the water, they also use it as ear drops or for domestic cleaning. Some of the locations identified in this study are Alam Damai, Sg Long, Sg Tekali, Taman Damai, Bukit Ampang and Ulu Yam. While one hot spring point was taken from Dusun Tua (just for external use, not for drink) and one ground water sample from Sg Gahal.

The objectives of this study were to determine Pb and Cd contents in these natural spring sources and commercialised bottled drinking water, and compare them with the standard.

MATERIALS AND METHODS

Alam Damai spring water sample was taken from an area with coordinate of 3°40'N 101°44'22"E. This water source is located along Persiaran Alam Damai near a 30-acre central park. Based on observation, there are housing areas located nearby together with a construction side about 40

meters downhill and 2 km uphill. However, there is abundance of forest reserve surrounding the water sources and no obvious contamination sources noted.

Second sampling was taken from Sungai Long (SILK highway) at coordinates of 3°0'8"N and 101°49'30"E. It is located near to the Saujana Impian exit (Exit16). Downhill to the area is a stone quarry about 5km and other nearest activity was 10km downhill, but none on uphill of the water source. Possible contamination may come from mineral sources as there was a presence of orange-colour tint along the water drainage.

Sungai Tekali spring water sample was taken from 3°6'35.3"N and 101°49'9.6"E location, that is about 8.3 km away from Semenyih Dam Reservoir. Here, there are few small and medium enterprise (SME) industries and housing areas nearby. During the sampling, authors noted the presence of garbages near the intake point. This possible source of contamination was most probably made by visitors who came and eat while waiting to fill their water bottles. Based on an interview with a local folk, who has been consumed the spring water for more than 40 years, claimed that the water quality today was differed from 10 years before.

Taman Damai, Batu 18 Hulu Langat spring water source available at 3°9'46.7"N and 101°50'47.6"E. It is a small town surrounded by forest, near the local police station. The water originated from uphill area. The resident collected the water in a tank that connected with pipe directly to nearby houses. Majority of Taman Damai residences uses both spring and treated water in their daily living activities.

Bukit Ampang spring water was sampled based on the coordinate of 3°7'42"N and 101°47'27"E. It is located near to the Taman Pinggiran Delima exit (B62). Here, there were various human activities nearby, but mainly recreational based. The nearest activity was 5km downhill. On the uphill area there was a sanctuary park present. Possible contamination may occur from animal excreta if activity is not properly managed and taken care of. During sampling, water source was not taken for consumption instead it as used for cleaning purposes as there were people took the water to wash car.

Ulu Yam spring water sample was taken from 3°36'22"N and 101°69'90"E. There are a hill and a forest reserved area. The only activity presence is a recreational park. It is located along Jalan Sungai Tua, in which is a main alternative road to the northern part of Malaysia. There is neither construction, nor a housing area within 5km

radius. Here, although there were 3 adjacent intake points, interestingly special one of them just used for topical eyes remedies.

National Youth Training Institute (IKBN) Dusun Tua, Hulu Langat was the only hot spring water sample taken. It was located at 3°13'69"N and 101°83'46"E. It's located near to the Jalan Langat main road that adjacent to a training institution. It's part of the Langat river basin area. There are houses and construction site about 500 meters uphill to the water source. Due to its location, possible contamination could come from the nearby major road and institutional facility.

The only ground water sampling was taken at Sungai Gahal, Hulu Langat at 3°12'26"N and 101°82'10"E. It's located at the side of the Jalan Langat main road near to a traditional village area. This water source is being used by the village people for the past 15 years and they have never used treated water supply since then. During the sampling the water was even used for preparation of food as the operator was a local hawker.

The study was conducted in 2014 as part of parcel of Environmental And Industrial Occupational Laboratory assignment. A field observation was carried out for water pH and temperature determination in situ. Data were collected using a pH meter, EC 10. The observation included water usage, canopy cover, and level of physical pollution in the surrounding area. Data such as longitude and latitude of sampling sites were determined using web based Global Positioning System mapping.

Water samples were collected in labeled plastic bottles. Bottles used for storage of samples were pre-washed in an acid bath with 10% nitric acid solution, followed by rinsing with de-ionized water, to minimize adsorption to bottle side wall. A total of 8 stations was selected. Samples were collected directly from the tubing into the pre-washed sample bottle and rinse with the sampling water in the field. Sample preservation was achieved by acidifying to pH less 4.0 with nitric acid (HNO₃). Samples were placed in a cooler with temperature conditions between 0 to 4°C and transport to a laboratory for analysis. For heavy metal analysis, samples were analysed by standard method for graphite furnace atomic absorption spectrophotometer (GFAAS) ⁷.

RESULTS

Pb and Cd concentrations were determined from 13 samples which consists of eight spring water points and five flavoured commercialised drinking water (Table 1). The result showed that Pb

concentrations in spring water points were between 1.8 and 37.3 ppb, while Cd concentrations ranged between 8.9 and 23.0 ppb.

Two out of eight spring water points have excessive Pb content than the drinking water standard of 10.0 ppb. The highest risk of Pb toxicity points were included the natural spring of Bukit Ampang and Ulu Yam points that used for cleaning purposes and eye drops respectively. Both points were also identified to contaminated with Cd at a very high concentration. And most of the

natural spring water has high Cd content exceeded the drinking water standard of 3.0 ppb. None of the natural spring water points selected in the study are safe.

Bottled flavoured commercialised drinking water samples are safer than the natural spring water. The amount of Pb ranges between 0.4 and 2.6 ppb, while the content of Cd ranges between 0.8 and 7.0 ppb. For sugar cane and lychee flavour have a high limit of Cd that exceeded the standard.

Table 1: Sample location and its uses with Lead and Cadmium content compared to the standard for drinking water

| No. | Sample | Uses | Lead (ppb) | Cadmium (ppb) |
|-----|-------------------|------------------|------------|---------------|
| 1 | Alam Damai | Drinking | 4.48 | 15.22* |
| 2 | Sungai Long | Drinking | 1.80 | 21.29* |
| 3 | Sungai Tekali | Drinking | 3.15 | 21.50* |
| 4 | Taman Damai | Food preparation | 8.33 | 21.87* |
| 5 | Bukit Ampang | Cleaning | 37.34* | 23.00* |
| 6 | Ulu Yam | Eye drop | 21.02* | 22.76* |
| 7 | Dusun Tua (HS) | Thermal bath | 9.78 | 13.31* |
| 8 | Sungai Gahal (GW) | Food preparation | 4.10 | 8.92* |
| 9 | Apple | Drinking | 0.37 | 0.92 |
| 10 | Black-currant | Drinking | 2.55 | 1.85 |
| 11 | Chrysanthemum tea | Drinking | 1.46 | 0.79 |
| 12 | Sugar cane | Drinking | 2.61 | 6.97* |
| 13 | Lychee | Drinking | 2.48 | 6.74* |

Source: The study data 2014 (HS: hot spring; GW: groundwater); *exceeded limit

DISCUSSION

Malaysian people are exposing themselves to health risk due to direct consumption of natural spring water points that all were found highly contaminated with heavy metals particularly sad and some of them with Pb. Similar findings were also found in other countries in their spring water⁸⁻¹⁰.

Cd was found to be high in all points. Those were samples taken located along the major roadside and almost rock and hilly areas that open to erosion. It may originate from the rocks as the natural type, of heavy metals in the spring water¹¹⁻¹². However, card also may present as the result of the road traffic and land status¹³⁻¹⁴. Pb in spring water was also found originated as natural or due to environmental contamination

Both Cd and Pb were found high in spring water points at Bukit Ampang and Ulu Yam. Bukit Ampang spring water was used to wash vehicles that exposed them through skin contact. Even though, Pb is not easily absorbed through the skin, but the exposure was found to be related to skin cancer¹⁵ and other skin disorder^{16,17,18}. On the other hand, spring water at Hulu Yam was used as

eye drops by passersby or local people. Both heavy metals were identified to be related with corneal oedema and scaring^{19,20}.

Other spring water points were found to contain high concentration of both metals and not suitable for daily consumption. The pollutions pose elevated risk for microbial contamination²¹, various cancers^{22,23}, and multiple diseases due to cellular oxidative stress effects of both²⁴.

Regarding bottled drinking water, sugar cane and lychee flavoured were found contaminated with cadmium. Excessive and prolonged consumption of these drinks will also pose many health ailments. Apple, black-currant and chrysanthemum tea flavoured drinking water were content very low level of both metals.

LIMITATION

This study has its limitations in that further supportive tests were not conducted during the water analysis study. We recommend that in the future, soil and sediment analysis, nutrient analysis as well as conductivity testing should be performed concurrently with the water analysis

testing so as to enable a more concrete explanation of the findings.

CONCLUSION

The results of this indicate that there are high levels of lead and cadmium in the natural waters within Klang Valley. These are very worrying, especially that these sources of water are constantly consumed directly almost every day for certain people in the area. Local authority needs to ban the activity until the metal content becomes less than the standard. Further tests, coupled with supportive soil and conductivity studies, are required to reconfirm these findings in the future. High content of metals in certain bottled drinking water granted for closer monitoring by the authority in ensuring the health of the nation.

ACKNOWLEDGEMENT

The authors would like to thank Puan Wan Aishah Bt Wan Nawawi with her Environmental Unit Team for their support in facilitating and conducting this study.

REFERENCES

1. United States Geological Survey. 2011. The water cycle: springs. U.S. Department of the Interior. Retrieved from <http://ga.water.usgs.gov/edu/watercycle.html>.
2. Naji A, Ismail A, Kamrani E, & Sohrabi T. 2014. Correlation of MT levels in livers and gills with heavy metals in wild tilapia (*Oreochromis mossambicus*) from the Klang River, Malaysia. *Bulletin Of Environmental Contamination And Toxicology* **92**(6):674-679.
3. Adiana G, Shazili N. A, Marinah M. A, & Bidai J. 2014. Effects of northeast monsoon on trace metal distribution in the South China Sea off Peninsular Malaysia. *Environmental Monitoring And Assessment* **186**(1):421-431.
4. Nagarajan R, Jonathan M. P, Roy P. D, Wai-Hwa L, Prasanna M. V, Sarkar S. K, & Navarrete-Lopez M. 2013. Metal concentrations in sediments from tourist beaches of Miri City, Sarawak, Malaysia (Borneo Island). *Marine Pollution Bulletin* **73**(1):369-373.
5. Mohd Adnan S. N, Yusoff S, & Piaw C. Y. 2013. Soil chemistry and pollution study of a closed landfill site at Ampar Tenang, Selangor, Malaysia. *Waste Management & Research* **31**(6):599-612.
6. Suratman S, Hang H. C, Shazili N. A, & Mohd Tahir, N. 2009. A preliminary study of the distribution of selected trace metals in the Besut River basin, Terengganu, Malaysia. *Bulletin Of Environmental Contamination And Toxicology* **82**(1):16-19.
7. Xingguang S, Meijia W, Yihua Z, Jiahua Z, Hanqi Z, & Qinhan J. 2003. Semi-online preconcentration of Cd, Mn and Pb on activated carbon for GFASS. *Talanta* **59**(5):989-997.
8. Walna B, & Siepak M. 2012. Heavy metals: their pathway from the ground, groundwater and springs to Lake Goreckie (Poland). *Environmental Monitoring And Assessment* **184**(5):3315-3340.
9. Virkutyte J, & Sillanpaa M. 2006. Chemical evaluation of potable water in Eastern Qinghai Province, China: human health aspects. *Environment International* **32**(1):80-86.
10. Aydin M. E, Ozcan S, & Ucar S. 2012. Heavy metal pollution of surface water sources of Konya Basin. *Metals and Related Substances in Drinking Water*:259-267.
11. Mikac I, Fiket Z, Terzic S, Baresic J, Mikac N, & Ahel M. 2011. Chemical indicators of anthropogenic impacts in sediments of the pristine karst lakes. *Chemosphere*. **84**(8):1140-1149.
12. Giusti L. 2011. Heavy metals in urban soils of Bristol (UK). Initial screening for contaminated land. *Journal of Soils and Sediments* **11**(8): 1385-1398.
13. Zou J, Dai W, Gong S, & Ma Z. 2015. Analysis of spatial variations and sources of heavy metals in farmland soils of Beijing suburbs. *PLoS One* **10**(2):e0118082.
14. Yan X. D, Gao D, Zhang F, Zeng C, Xiang W, & Zhang M. 2013. Relationships between Heavy Metal Concentrations in Roadside Topsoil and Distance to Road Edge Based on Field Observations in the Qinghai-Tibet Plateau, China. *International Journal of Environmental Research and Public Health* **10**(3):762-775.

15. Rashidi M, Rameshat MH, Gharib H, Rouzbahani R, Ghias M, & Poursafa P. 2012. The association between spatial distribution of common malignancies and soil lead concentration in Isfahan, Iran. *Journal of Research in Medical Sciences* 17(4): 348-354.
16. Afridi HI, Kazi TG, Kazi N, Kandhro GA, Baig JA, Shah AQ, Khan S, Kolachi NF, Wadhwa SK, Shah F, Jamali MK, & Arain MB. 2011. Evaluation of cadmium, chromium, nickel, and zinc in biological samples of psoriasis patients living in Pakistani cement factory area. *Biological Trace Element Research* 142(3):284-301.
17. Onishi K, Otani S, Yoshida A, Mu H, & Kurozawa Y. 2015. Adverse Health Effects Of Asian Dust Particles And Heavy Metals In Japan. *Asia-Pacific Journal Of Public Health* 27(2):NP1719-1726.
18. Kang Y, Yin Y, Zhang QY, Li LS, Zeng LX, Luo JW, & Wong MH. 2014. Heavy metal influence on BDE-47 uptake in the human KERTr keratinocyte cell line. *Environmental Toxicology* 29(3):354-361.
19. Amry MA, Al-Saikhan F, & Ayoubi A. 2011. Toxic effect of Cadmium found in eyeliner to the eye of a 21 year old Saudi woman: A case report. *Saudi Pharmaceutical Journal* 19(4):269-272.
20. Weidner WJ, Waddell DS, & Sillman AJ. 2000. Low levels of cadmium chloride alter the immunoprecipitation of corneal cadherin-complex proteins. *Archives Of Toxicology* 74(10):578-581.
21. Zhang J, Wang LH, Yang JC, Liu H, & Dai JL. 2015. Health risk to residents and stimulation to inherent bacteria of various heavy metals in soil. *The Science Of The Total Environment* 508:29-36.
22. Lu SY, Zhang HM, Sojinu SO, Liu GH, Zhang JQ, & Ni HG. 2015. Trace elements contamination and human health risk assessment in drinking water from Shenzhen, China. *Environmental Monitoring And Assessment* 187(1):4220.
23. Golasik M, Jawien W, Przybylowicz A, Szyfter W, Herman M, Golusinski W, Florek E, & Piekoszewski W. 2015. Classification models based on the level of metals in hair and nails of laryngeal cancer patients: diagnosis support or rather speculation? *Metallomics : Integrated Biometal Science* 7(3):455-465.
24. Bibi M, Hashmi MZ, & Malik RN. 2015. The level and distribution of heavy metals and changes in oxidative stress indices in humans from Lahore district, Pakistan. *Human & Experimental Toxicology* pii: 0960327115578063. [Epub ahead of print].